

## Nitrous Oxide Emission from Nitrogen Saturated Forest

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Nitrogen deposition into the forest ecosystems is recently increased by human activities. Although forests are generally nitrogen limited, it is reported that some forests are nitrogen saturated by the increase of such nitrogen deposition<sup>1)</sup>. Increasing in the nitrogen deposition can affect the production and consumption of nitrous oxide (N<sub>2</sub>O), which is one of the greenhouse gases and is produced by nitrification and denitrification in soils. However, how the increase of nitrogen deposition affects N<sub>2</sub>O production by nitrification and denitrification is not clear. So, the purpose of this study is to examine the processes of N<sub>2</sub>O production in nitrogen saturated forest's soil.

Study site was the Field Museum Tama (FM Tama) of the Tokyo University of Agriculture and Technology, Hachioji, Tokyo. The FM Tama had been reported as the nitrogen saturated forest<sup>2)</sup>. Soil samples were collected from 0-15cm depth. The soil was brown forest soil. Soils were sieved (2mm) and its moisture content was adjusted to 60% of water-holding capacity. Samples were placed in 300ml glass bottles, and a rubber stopper with sampling ports was placed each bottle. Acetylene (C<sub>2</sub>H<sub>2</sub>) was injected into bottles so that the headspace partial pressure is 10kPa, 10Pa, or no C<sub>2</sub>H<sub>2</sub>. The bottles were incubated in the dark for 10 days at 25 degrees Celsius. Headspace was sampled every two or four days to measure N<sub>2</sub>O concentrations by GC-ECD. The difference between N<sub>2</sub>O production rates with the three treatments (10kPa, 10Pa and control) can be used to calculate the N<sub>2</sub>O produced by nitrification, denitrification and N<sub>2</sub>O reduction<sup>3)</sup>.

The difference between N<sub>2</sub>O concentration of control and that under 10Pa treatment was not statistically significant and N<sub>2</sub>O concentration of 10kPa treatment was statistically higher than other treatments. The production rate of N<sub>2</sub>O was largest under 10kPa, and control samples showed the smallest rates (236.4 and 2.8[micromol N<sub>2</sub>O/kg fresh soil/h], respectively). Production of N<sub>2</sub>O was calculated as follows; nitrification N<sub>2</sub>O: -52.1micromol N<sub>2</sub>O/kg fresh soil/h, denitrification N<sub>2</sub>O: 54.9micromol N<sub>2</sub>O/kg fresh soil/h, denitrification N<sub>2</sub>: 181.5micromol N<sub>2</sub>O/kg fresh soil/h. N<sub>2</sub>O production rate by nitrification was calculated as negative rate, but the difference between N<sub>2</sub>O concentration of control and that of 10Pa treatment was not statistically significant, and we treated this minus value as zero. Therefore, it was suggested that N<sub>2</sub>O was not emitted by nitrification and N<sub>2</sub>O was mainly emitted by denitrification. Moreover, there was significant production of N<sub>2</sub> by denitrification exceeded total N<sub>2</sub>O production. Thus, it was concluded that the processes of N<sub>2</sub>O emission were mainly dominated by the reduction of N<sub>2</sub>O by denitrification at the study site, though the soils were in aerobic condition.

1) Aber, J. *et al.* (1989), *Bioscience* **39**, 378-386

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3) Davidson, E.A. *et al.* (1986), *Appl. Environ. Microbiol.* **52**, 1280-1286