

## Temperature-dependent thermal expansivities of aluminum-free silicate melts and borosilicate melts

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Thermal expansivities ( $dV/dT$ ) of silicate melts are essential in a thermodynamic calculation of phase equilibria in magmatic system as a function of pressure and temperature and in a numerical simulation of flow and thermal structures in glass melting furnace. Previous studies have been suggested that the  $dV/dT$  of aluminosilicate melts (Lange, 1996; Potuzak et al., 2006) and magmatic silicate melts (Lange, 1997; Ghiorso and Kress, 2004) is a function of composition, but independent of temperature. On the other hand, it has been reported that the  $dV/dT$  of  $\text{SiO}_2$ - $\text{TiO}_2$ - $\text{Na}_2\text{O}$  melt (Liu and Lange, 2001) and  $50\text{SiO}_2$ - $25\text{CaO}$ - $25\text{MgO}$  melt (Gottsmann and Dingwell, 2000) decrease with increasing temperature. Recently, we found that simulated-radioactive waste glass melt which has sodium-borosilicate composition also shows negative temperature-dependent  $dV/dT$  (Sugawara et al., 2013). We carried out density measurements for sodium-silicate melts ( $(100-x)\text{SiO}_2$ - $x\text{Na}_2\text{O}$ ,  $x=23$  or  $32.3$  mol%), commercial soda-lime silicate melt ( $71\text{SiO}_2$ - $6\text{MgO}$ - $9\text{CaO}$ - $14\text{Na}_2\text{O}$ , mol%) and borosilicate melts ( $66.6\text{SiO}_2$ - $y\text{B}_2\text{O}_3$ - $(33.33-y)\text{Na}_2\text{O}$  where  $y=8.3, 16.6, 25$ ;  $66.6\text{SiO}_2$ - $(12.5+z)\text{B}_2\text{O}_3$ - $(4.2-z)\text{Al}_2\text{O}_3$ - $z\text{CaO}$ - $(16.7-z)\text{Na}_2\text{O}$  where  $z=0$  or  $4.2$  mol%). The temperature and compositional dependences of the  $dV/dT$  are discussed based on the new density data and the literature data.

The high-temperature density (dHT) measurement has been made by double-bob Archimedean method between 1173K and 1665K. The glass samples were annealed around glass transition temperature ( $T_g$ ) for 6-396 hours and quenched. Then the density of annealed glasses at 298K (d298) and linear thermal expansivity (dL/L) were determined by Archimedean method and TMA, respectively. The densities of supercooled melt around  $T_g$  (dTg) were calculated from the d298 and the dL/L of glasses. Then, molar volume as a function of temperature and the  $dV/dT$  of melts were obtained by combining the dTg and the dHT.

The  $dV/dT$  values of all samples examined in this study show negative temperature dependence. In the sodium silicate melts, the temperature dependence of the  $dV/dT$  is remarkable when the  $\text{SiO}_2$  content increases from 50 to 67 mol%, while the  $dV/dT$  becomes close to zero as further increase in the  $\text{SiO}_2$  content. The negative temperature-dependent  $dV/dT$  observed in the  $71\text{SiO}_2$ - $6\text{MgO}$ - $9\text{CaO}$ - $14\text{Na}_2\text{O}$  melt can be reproduced by an additive sum of the  $dV/dT$  of  $67.8\text{SiO}_2$ - $32.2\text{Na}_2\text{O}$ , diopside (Gottsmann and Dingwell, 2000) and wollastonite (Potuzak et al., 2006) melts. High-temperature Raman spectroscopy for the  $\text{SiO}_2$ - $\text{Na}_2\text{O}$  and  $\text{SiO}_2$ - $\text{Na}_2\text{O}$ - $\text{MgO}$  melts has been indicated that amount of Q4 species increases with increasing temperature and  $\text{SiO}_2$  and  $\text{MgO}$  contents (Maehara et al., 2004, 2005). Therefore, the temperature dependent  $dV/dT$  for the sodium-silicate, commercial soda-lime silicate and diopside melts can be rationalized by an increase of rigid Q4 species at high temperature. The temperature dependence of the  $dV/dT$  is most remarkable in the  $66.6\text{SiO}_2$ - $8.3\text{B}_2\text{O}_3$ - $25\text{Na}_2\text{O}$  melt among the borosilicate melts. The  $dV/dT$  decreases with replacement of  $\text{Na}_2\text{O}$  by  $\text{B}_2\text{O}_3$  or  $\text{CaO}$  and of  $\text{B}_2\text{O}_3$  by  $\text{Al}_2\text{O}_3$ , suggesting that partial molar  $dV/dT$  of  $\text{B}_2\text{O}_3$  depends on temperature-induced coordination change of boron and their composition dependence (Wu and Stebbins, 2010).

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