Gb-021

Room: IM

Al2O3 solubility in orthopyroxene coexisting with spinel + quartz and sapphirine + quartz in the Mg3Al2Si3O12 system

Toshisuke Kawasaki [1]

[1] Earth Sci., Ehime Univ.

High pressure experiments were carried out to determine Al2O3 solubility in orthopyroxene in equilibrium with assemblages of spinel + quartz and sapphirine + quartz in the Mg3Al2Si3O12 (pyrope composition) system at pressures 0.7 to 2.2 GPa and temperatures 1100C to 1400C.

The phase boundary of the univariant reaction: 2MgAl2O4 (Spl) + SiO2 (Qtz) = Mg2Al4SiO10 (Spr) locates within a certain pressure between 1.5 and 1.75 GPa at 1300C. Solubility of Al2O3 in orthopyroxene increases with increasing temperature and decreases with pressure. Orthopyroxene behaves as Mg2Si2O6-MgAl2SiO6 binary regular solution.

High pressure experiments were carried out to determine Al2O3 solubility in orthopyroxene in equilibrium with mineral assemblages of spinel + quartz and sapphirine + quartz: MgAl2SiO6 (Opx) = MgAl2O4 (Spl) + SiO2 (Qtz), and 2MgAl2SiO6(Opx) = Mg2Al4SiO10 (Spr) + SiO2 (Qtz) in the Mg3Al2Si3O12 (pyrope composition) system at pressures 0.7 to 2.2 GPa and temperatures 1100C to 1400C. Run times were 2-270 hours.

Starting materials used in this study were the mineral mixture of olivine + cordierite + spinel sintered at 1 atm and 1300C, pyrope garnet synthesized at 3.0 GPa and 1300C, and glass. Pulverized materials (about 5-10 microns) were put into graphite capsules in the BN sleeve, which was situated in the talc+Pyrex glass pressure transmitting medium. All experiments were conducted under dry condition. Pressure calibrations were carried out by the quartz-coesite transformation at 1000C, and Bi(1)-Bi(2) transition at room temperature. Temperatures were monitored by Pt-Pt13%Rh thermocoupple.

The assemblage of orthopyroxene + spinel + quartz was stable at 1.5 GPa and 1300C. At 1.75 GPa, the stable assemblage changed to orthopyroxene + sapphirine + quartz. This indicates that the phase boundary of the univariant reaction: 2MgAl2O4 (Spl) + SiO2 (Qtz) = Mg2Al4SiO10 (Spr) locates within 1.5 and 1.75 GPa at 1300C. Solubility of Al2O3 in orthopyroxene increases with increasing temperature and decreases with pressure. Orthopyroxene behaves as Mg2Si2O6-MgAl2SiO6 binary regular solution.