

## Electron spin resonance study of free radicals in amorphous ice

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H<sub>2</sub>O are deposited on solar system bodies of ice and become amorphous not crystalline ice. Since amorphous ice can adsorb many kinds of molecules because of its porous structure, photolysis of adsorbed molecules is important to know chemical evolutions in space. Electron spin resonance (ESR) method can directly detect free radicals, which are very important in chemical reactions. Amorphous ice samples were deposited on the copper finger at 77 K in a vacuum from water vapor and irradiated by low-pressure mercury lamp. The ESR spectra of samples doped with 1% H<sub>2</sub>O<sub>2</sub> showed the radical formation of HO<sub>2</sub> and we identified the matrix as amorphous ice by comparing with spectra reported by Bednarek (1996). We report the ESR analysis of UV-irradiated amorphous ice samples adsorbing simple molecules.