

## Pressure-temperature relationship of CO<sub>2</sub> species in the H<sub>2</sub>O-CO<sub>2</sub> system

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Fluid inclusions are classified morphologically into 4 categories; liquid inclusion, gaseous inclusion, polyphase inclusion and CO<sub>2</sub>-bearing inclusion. Liquid CO<sub>2</sub> phase is visible on the microscope at room temperature in the case of high-CO<sub>2</sub> fluid inclusions. Liquid CO<sub>2</sub> phase occurs surrounding a gas bubble in a fluid inclusion. The liquid CO<sub>2</sub> converts to CO<sub>2</sub> hydrate with decreasing temperatures. The transition temperatures from gas CO<sub>2</sub> to liquid CO<sub>2</sub>, and from liquid CO<sub>2</sub> to CO<sub>2</sub> hydrate vary between 9.8 and 28.3 degree C, and 2.2 and 10.4 degree C, respectively. The NaCl concentration in fluid inclusions vary from 4 to 6 wt% approximately. A CO<sub>2</sub> concentration in a fluid inclusion can be estimated from a volume of liquid CO<sub>2</sub> in a fluid inclusion. By relating the NaCl and CO<sub>2</sub> contents to gas-liquid CO<sub>2</sub> transition and liquid-hydrate CO<sub>2</sub> transition temperatures, the almost all transition temperatures are identified to be closely associated to NaCl concentration.