

## Determination of triple oxygen isotopic compositions of atmospheric nitrous oxide by using continuous-flow isotope ratio MS

# Daisuke Komatsu[1]; satoru ohkubo[1]; Toyoho Ishimura[2]; Fumiko Nakagawa[3]; Urumu Tsunogai[3]

[1] Earth and Planetary Sci., Hokkaido Univ.; [2] Natural History Sciences, Hokkaido University; [3] Earth & Planetary Sci., Hokkaido Univ.

The triple oxygen isotopic compositions ( $^{18}\text{O}/^{16}\text{O}$  and  $^{17}\text{O}/^{16}\text{O}$ ) of nitrous oxide ( $\text{N}_2\text{O}$ ) in atmosphere can be a useful tracer to clarify their sources and N cycles in atmosphere. The triple oxygen isotopic compositions of  $\text{N}_2\text{O}$  have been usually determined by using conventional IRMS system using  $\text{O}_2$  molecule converted from  $\text{N}_2\text{O}$  through multiple reaction/purification steps. The traditional methods, however, required at least 1-100  $10^3$  nmol quantities of  $\text{N}_2\text{O}$  so that applications of the methods to various environmental  $\text{N}_2\text{O}$  samples were difficult. Thus, we developed a rapid and sensitive analytical system to determine the triple oxygen isotopic compositions of  $\text{N}_2\text{O}$  in nmol quantities using continuous-flow IRMS (CF-IRMS) without the cumbersome and time-consuming pretreatments. We cannot determine  $^{17}\text{O}/^{16}\text{O}$  ratio of  $\text{N}_2\text{O}$  directly by measuring the masses 44, 45, and 46 of  $\text{N}_2\text{O}$  introduced to IRMS, because the measured output of mass 45 from IRMS consists of  $^{14}\text{N}^{15}\text{N}^{16}\text{O}$ ,  $^{15}\text{N}^{14}\text{N}^{16}\text{O}$ , and  $^{14}\text{N}^{14}\text{N}^{17}\text{O}$ . Thus, addition to the  $\text{N}_2\text{O}$  isotopic analysis at the masses 44, 45, and 46, the  $^{15}\text{N}/^{14}\text{N}$  ratio is determined separately for the same sample  $\text{N}_2\text{O}$ . To attain this purpose, two instrumental approaches were done. In the first system, the  $\text{N}_2^+$  fragment ion beams of  $\text{N}_2\text{O}$  at masses 28 and 29 were used to determine the  $^{15}\text{N}/^{14}\text{N}$  ratio of  $\text{N}_2\text{O}$ . While the analytical precisions better than 0.5 per mil for 20 nmol  $\text{N}_2\text{O}$  injections and better than 1.0 per mil for 7 nmol  $\text{N}_2\text{O}$  injections were obtained for  $^{15}\text{N}/^{14}\text{N}$  ratio, we found that the accuracy strongly depended on the quantities introduced. In the second system, the  $\text{N}_2$  molecules, converted from  $\text{N}_2\text{O}$  using an on line Cu reduction furnace (720 degree) was used to determine the  $^{15}\text{N}/^{14}\text{N}$  ratio of  $\text{N}_2\text{O}$ . The analytical precisions better than 0.1 per mil for 5 nmol  $\text{N}_2\text{O}$  injections and better than 0.4 per mil for 1 nmol  $\text{N}_2\text{O}$  injections were obtained for  $^{15}\text{N}/^{14}\text{N}$  ratio with stable accuracy irrespective of the quantities introduced. Thus, we introduced  $\text{N}_2$  and  $\text{N}_2\text{O}$  to CF-IRMS repeatedly by using an automatic multi-injection system to improve analytical precision statistically, so that the analytical precisions in the  $^{17}\text{O}/^{16}\text{O}$  ratio of  $\text{N}_2\text{O}$  and nitrate better than 1.0 per mil for 100 nmol injections and better than 4.0 per mil for 20 nmol injections were obtained. Besides to  $\text{N}_2\text{O}$ , we can also apply the method to determine the triple oxygen isotopic composition of nitrate based on the isotopic analysis of  $\text{N}_2\text{O}$  quantitatively converted from nitrate through the simple reactions using spongy cadmium and sodium azide in an acetic acid buffer. We present the triple oxygen isotopic compositions of  $\text{N}_2\text{O}$  and nitrate in atmosphere determined by using this system.