

Rapid isotope analysis of the deuterium/hydrogen ratio of hydrogen in atmosphere at several hundreds ppm levels

Keita Yamada[1]; Mariko Hayashida[2]; Sakae Toyoda[3]; Naohiro Yoshida[4]

[1] Environ. Chem. and Engr, Tokyo Tech.

; [2] Env. Sci. Tech., Tokyo Tech.; [3] Environmental Chemistry and Engineering, Tokyo Tech; [4] IGSSE, Tokyo Institute of Technology

Molecular hydrogen (H_2) is the second most abundant trace gas in the atmosphere (after methane, CH_4), with a mixing ratio of ca. 500 ppb (e.g., Novelli et al., 1999). In the near future, there is a possibility for a change in a balance of anthropogenic emissions of H_2 and an increase overall because it is considered to be an important future alternative to fossil fuels. The increase in atmospheric H_2 levels leads to an important change in air quality, such as an increase of the lifetimes of other important trace gases such as CH_4 and carbon monoxide (CO) in the troposphere or a change in the water budget in the stratosphere. To identify the cause of the increase of the atmospheric H_2 levels and its influence on atmospheric chemistry, it is important to constrain the current budget of atmospheric H_2 .

Measurements of stable isotope ratio (i.e., D/H ratio) in atmospheric H_2 and individual H_2 source and sink processes have the potential to help constrain the local, regional and global H_2 budget. However, the measurement of the D/H ratio in atmospheric H_2 and its sources has so far been less developed, because of difficulties in the preparation of samples that is required before analysis by a dual-inlet isotope ratio mass spectrometer (IRMS) used by previous studies (e.g., Ehhalt, 1966; Friedman and Scholz, 1974; Gerst and Quay, 2000).

More recently, an off-line H_2 gas separation from the air matrix followed by introduction to an IRMS (Rahn et al., 2002) and an on-line H_2 gas separation and concentration coupled with a gas chromatograph (GC) - IRMS (Rhee et al., 2004) have been developed. These techniques not only reduced the amount of air required for isotopic analysis of atmospheric H_2 by a order of three compared to the previous studies, but also obviate the need for catalytic conversion of H_2 . However, these techniques require the vacuum line system equipped with a cold-trap for concentration of H_2 gas operated at very low temperature (30 - ca.60 Kelvin).

In this study we tried a preliminary development of a simple method that avoids the use of the cold-trap for concentration of H_2 gas. The method used GC-IRMS system (Finnigan Delta plus-XL coupled with HP6890 GC via GC-III interface, ThermoFinnigan). The GC was equipped with a split/splitless injector and capillary column (CARBOXENTM 1010PLOT, 30m, Supelco). Top part of the capillary column (ca.20 cm) immersed in 77K liquid N_2 . Up to 1 mL of air sample is injected into the injector of the GC with splitless mode by a gas-tight syringe. Using this method, the D/H ratio of atmospheric H_2 at several hundreds ppm levels was determined with precision of 2 permil. The method is useful for atmospheric samples with relatively high H_2 concentrations, such as biomass burning source.