

Geochemical water cycle and evolution of mantle redox state

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A numerical model is constructed to simulate the evolution of gas composition and flux degassing from mantle and that of mantle redox state through geochemical water cycle.

For the evolution of the Earth's atmospheric composition, plausible estimates have been done about CO₂ and O₂ so far (Kasting, 1993). However, other gas species such as H₂, N₂, NH₃, CH₄ remain to be studied. It is important to clarify the evolution of atmospheric composition, particularly redox state, for understanding not only the evolution of surface environment of the Earth but also the origin and evolution of life. There exist, however, many uncertainties especially for the early Earth.

The redox state of the atmosphere is controlled by the balance of supply and loss of oxidized and reduced molecules for the atmosphere. Therefore, this study focuses on H and O which are the most abundant volatile elements and the numerical model of H₂ and H₂O degassing from mantle to atmosphere-ocean system is formulated taking into account the change in mantle composition due to redox reaction.

In our model, degassing occurs from mid-oceanic ridge where magma production rate is the highest on the Earth. The ratio of H₂O and H₂ in degassing gas is obtained from the chemical equilibrium calculation between gas and magma generated by partial melting of mantle. Here we consider the reaction $\text{H}_2\text{O} + 2\text{FeO} = \text{H}_2 + \text{Fe}_2\text{O}_3$. A realistic upper mantle composition is given for model mantle while the present concentrations of Fe₂O₃ and H₂O in the mantle are given by observational constraints. The magma production rate is assumed to be equal to the oceanic crust production rate which changes with mantle temperature. Supposing that the volume of seawater is constant (Kasting and Holm, 1992), H₂O inflow flux to mantle is balanced with H₂O degassing flux from mantle. In this case, the secular change of H₂O concentration in mantle is equivalent to the integrated amount of degassing H₂.

The H₂ / H₂O ratio of degassing gas rises as the concentration of Fe₂O₃ in mantle is lower. The estimate of Fe₂O₃ concentration in mantle includes uncertainty. In the case of present Fe₂O₃ concentration at 0.1 wt% , which is near the lower estimate from observation, its concentration was as low as 0.04 wt% 4 billion years ago. For 4 billion years, the H₂ degassing flux decreases from 6.5×10^{12} mol/yr to the present value 7.8×10^{11} mol/yr. During this period, 0.4 ocean mass of H₂O was lost from mantle. In the case of present Fe₂O₃ concentration at 0.4 wt% , which is near the upper estimate from observation, its concentration was as low as 0.38 wt% 4 billion years ago. The H₂ degassing flux decreases from 2.7×10^{12} mol/yr to the present value 0.95×10^{11} mol/yr. In this case, 0.15 ocean mass of H₂O was lost from mantle for 4 billion years.

These results suggest that the degassing flux of H₂ was 10 - 30 times larger and that atmospheric H₂ concentration might be as high as 2 - 50 vol% 4 billion years ago when the degassing flux is balanced with H₂ escape flux from atmosphere to space (Tian et al., 2005). Hence, the Earth's early atmosphere might have had significantly reduced composition.