

In situ detection of reactive nitrogen species using thermal dissociation laser-induced fluorescence in Tokyo

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In the troposphere, nitrogen oxides ($\text{NO}_x = \text{NO}$ and NO_2) are mainly emitted from artificial activities such as burning fossil fuel. NO_x reacts with OH and RO_2 radicals in the troposphere and is converted to reactive nitrogen species, such as peroxy nitrates, alkyl nitrates, and HNO_3 . These compounds have an effect on the global distribution of NO_x in the troposphere because insoluble compounds, such as PAN, can be transported over long distances and then release NO_x again by thermal decomposition and photochemical reaction. HNO_3 is one of the major sinks of nitrogen oxides to control the lifetime of it. For these reasons, it is necessary to know the concentrations of not only NO_x but also these reactive nitrogen species in the troposphere. However, most of the measurements, which are used today to detect these compounds, have the potential for sample loss because of using a filter or gas chromatography for collection, and continuous rapid measurement is difficult.

In this work, we have developed a four channel thermal dissociation laser-induced fluorescent (TD-LIF) instrument to measure NO_2 , total peroxy nitrates, total alkyl nitrates, and HNO_3 separately in real time [1]. These compounds thermally dissociate to yield NO_2 at different temperature. An ambient sample flows rapidly through three ovens where dissociations of peroxy nitrates, alkyl nitrates, and HNO_3 to NO_2 occur. The four different temperatures, 25 (room temperature), 200, 430, and 650°C were used to observe NO_2 , peroxy nitrates, alkyl nitrates, and HNO_3 , separately. After passing through the ovens, the NO_2 was detected using LIF system at 445 nm. The NO_2 LIF signal is the sum of the NO_2 contained in all compounds that dissociated at or below the oven temperature, and the difference between the NO_2 observed in separate channels is associated with a specific class of compounds. The TD-LIF system is expected to be able to identify the interferences of reactive nitrogen species, which cause the overestimation of NO_2 in the chemiluminescence technique using a molybdenum converter (Mo-CL) [2].

The instrument was applied to the field measurements in central Tokyo during summer 2009. In this presentation, the performance of the TD-LIF instrument and the results of the field measurements will be reported. Additionally, we will discuss the interference of reactive nitrogen species in the NO_2 concentration determinations by the Mo-CL technique by comparing NO_2 concentrations measured using the TD-LIF and Mo-CL systems.

References

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