

SMP057-04

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熱重量測定によるCa(OH)₂とCa(OD)₂の脱水反応の速度論的解析

Kinetic study on dehydration reactions of Ca(OH)₂ and Ca(OD)₂ by Thermo Gravimetry measurements

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Hydrous minerals are expected to be present in the downgoing slabs and the uppermost earth's mantle. Knowledge of the physical properties of hydrous minerals, especially, mechanism of dehydration process is important to understand the source of water in the mantle. Portlandite, Ca(OH)₂, has a CdI₂ type structure (trigonal, space group P-3m1) and one of the most simple hydrous minerals. Ca(OD)₂ is often used for neutron scattering experiments instead of Ca(OH)₂ because of the incoherent cross section of H. However, it is not so clear whether Ca(OD)₂ is good analogous with Ca(OH)₂ in terms of some physical properties.

In order to understand kinetics of dehydration reactions of Ca(OH)₂ and Ca(OD)₂, we performed Thermo Gravimetry (TG) measurements at several different rate of elevating temperature (1, 5, 10 and 20 K/min) and determined activation energies of dehydration reactions by the Ozawa method (e.g., Ozawa, 2005). During TG measurements Ar gas was introduced in to the TG furnace to avoid the reaction between the samples and CO₂. We obtained activation energies of 16.6 and 15.2 kcal/mol for Ca(OH)₂ and Ca(OD)₂, respectively. When the activation energy of dehydration reaction reflects the bonding energy of O-H(D), the activation energy for Ca(OD)₂ is expected to be significantly larger than that for Ca(OH)₂. Therefore, the activation process during the dehydration reaction should be different from the breaking of O-H(D) bonding.

We also measured the FT-IR spectra of Ca(OH)₂ and Ca(OD)₂ at elevated temperatures. The positions of OH⁻ and OD⁻ stretching bands are shifted to lower wavenumbers with elevating temperature. The temperature coefficients for those bands are -0.28(4) and -0.43(3) cm⁻¹/K, respectively. It is interesting that both of those bands disappear at about 543 K which is more than 100 K below the onset temperature of dehydration.

キーワード: 含水鉱物, 脱水反応, 熱重量測定, 速度論的解析, Ca(OH)₂, Ca(OD)₂

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