

## Temperature-dependent thermal expansivities of aluminum-free silicate melts and borosilicate melts

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Thermal expansivities ( $dV/dT$ ) of silicate melts are essential in a thermodynamic calculation of phase equilibria in magmatic system as a function of pressure and temperature and in a numerical simulation of flow and thermal structures in glass melting furnace. Previous studies have been suggested that the  $dV/dT$  of aluminosilicate melts (Lange, 1996; Potuzak et al., 2006) and magmatic silicate melts (Lange, 1997; Ghiorso and Kress, 2004) is a function of composition, but independent of temperature. On the other hand, it has been reported that the  $dV/dT$  of  $SiO_2$ - $TiO_2$ - $Na_2O$  melt (Liu and Lange, 2001) and  $50SiO_2$ - $25CaO$ - $25MgO$  melt (Gottsmann and Dingwell, 2000) decrease with increasing temperature. Recently, we found that simulated-radioactive waste glass melt which has sodium-borosilicate composition also shows negative temperature-dependent  $dV/dT$  (Sugawara et al., 2013). We carried out density measurements for sodium-silicate melts ( $(100-x)SiO_2$ - $xNa_2O$ ,  $x=23$  or  $32.3$  mol%), commercial soda-lime silicate melt ( $71SiO_2$ - $6MgO$ - $9CaO$ - $14Na_2O$ , mol%) and borosilicate melts ( $66.6SiO_2$ - $yB_2O_3$ - $(33.33-y)Na_2O$  where  $y=8.3, 16.6, 25$ ;  $66.6SiO_2$ - $(12.5+z)B_2O_3$ - $(4.2-z)Al_2O_3$ - $zCaO$ - $(16.7-z)Na_2O$  where  $z=0$  or  $4.2$  mol%). The temperature and compositional dependences of the  $dV/dT$  are discussed based on the new density data and the literature data.

The high-temperature density (dHT) measurement has been made by double-bob Archimedean method between 1173K and 1665K. The glass samples were annealed around glass transition temperature ( $T_g$ ) for 6-396 hours and quenched. Then the density of annealed glasses at 298K (d298) and linear thermal expansivity (dL/L) were determined by Archimedean method and TMA, respectively. The densities of supercooled melt around  $T_g$  (dTg) were calculated from the d298 and the dL/L of glasses. Then, molar volume as a function of temperature and the  $dV/dT$  of melts were obtained by combining the dTg and the dHT.

The  $dV/dT$  values of all samples examined in this study show negative temperature dependence. In the sodium silicate melts, the temperature dependence of the  $dV/dT$  is remarkable when the  $SiO_2$  content increases from 50 to 67 mol%, while the  $dV/dT$  becomes close to zero as further increase in the  $SiO_2$  content. The negative temperature-dependent  $dV/dT$  observed in the  $71SiO_2$ - $6MgO$ - $9CaO$ - $14Na_2O$  melt can be reproduced by an additive sum of the  $dV/dT$  of  $67.8SiO_2$ - $32.2Na_2O$ , diopside (Gottsmann and Dingwell, 2000) and wollastonite (Potuzak et al., 2006) melts. High-temperature Raman spectroscopy for the  $SiO_2$ - $Na_2O$  and  $SiO_2$ - $Na_2O$ - $MgO$  melts has been indicated that amount of Q4 species increases with increasing temperature and  $SiO_2$  and  $MgO$  contents (Maehara et al., 2004, 2005). Therefore, the temperature dependent  $dV/dT$  for the sodium-silicate, commercial soda-lime silicate and diopside melts can be rationalized by an increase of rigid Q4 species at high temperature. The temperature dependence of the  $dV/dT$  is most remarkable in the  $66.6SiO_2$ - $8.3B_2O_3$ - $25Na_2O$  melt among the borosilicate melts. The  $dV/dT$  decreases with replacement of  $Na_2O$  by  $B_2O_3$  or  $CaO$  and of  $B_2O_3$  by  $Al_2O_3$ , suggesting that partial molar  $dV/dT$  of  $B_2O_3$  depends on temperature-induced coordination change of boron and their composition dependence (Wu and Stebbins, 2010).

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